

# Ductility in Metals

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2016 Summer Engineering Seminar

Post 1982 Penny



97.5 wt% Zn

Pre 1982 Penny

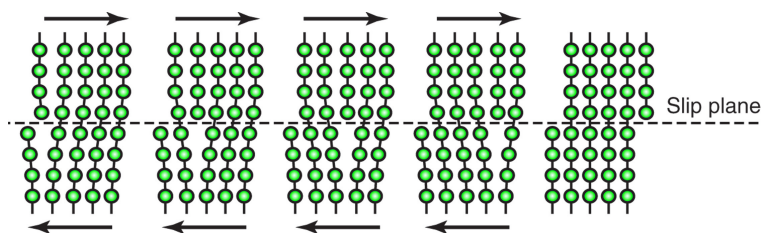
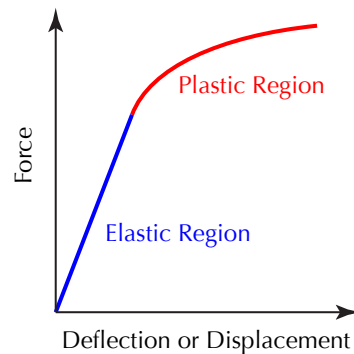


95 wt% Cu

## DEFORMATION OF METALS:

- Elastic Region: stretching of atomic bonds (similar to springs)

- Plastic Region: permanent deformation...WHY???
- Dislocations

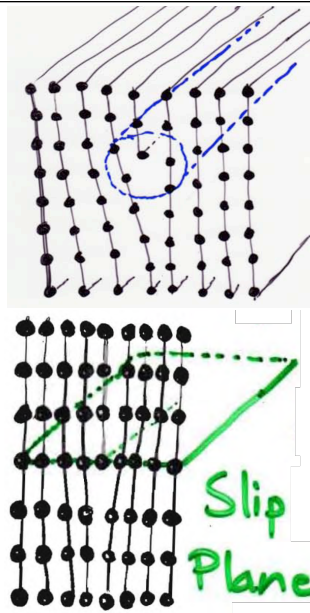


## DISLOCATION STRUCTURE:

- A dislocation extends through the material in three dimensions (we can think of it as a line or curve meandering throughout the crystal).

- Dislocation motion (*a.k.a. slip*) occurs more easily in metals compared to covalent solids (e.g., diamond) or ionic solids (e.g., NaCl).

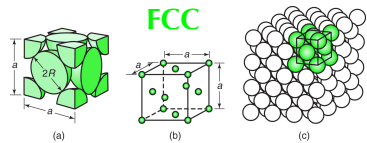
- For “easy” slip want atoms close together in the slip direction, but with large separation perpendicular to the slip plane.



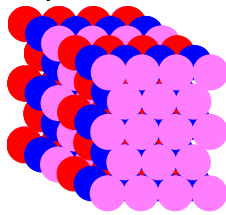
## CLOSE-PACKED STRUCTURES:

Face-Centered Cubic

FCC

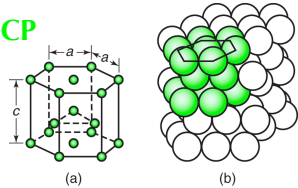


Layer: **A**BCABC...

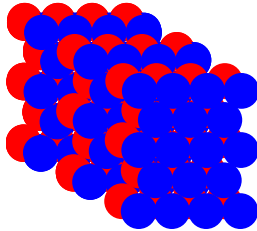


Hexagonal Close Packed

HCP



Layer: **A**BABAB...



Post 1982 Penny



Zn is HCP

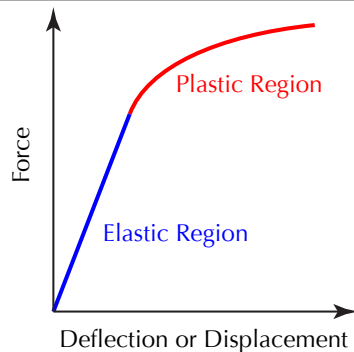
Pre 1982 Penny



Cu is FCC

## STRAIN HARDENING IN METALS:

- Higher force needed for continued plastic deformation.
- Dislocation movement becomes more difficult as deformation proceeds.
- Dislocations get “tangled” up with one another and it becomes harder for them to move, so more force is required to continue deforming the metal.



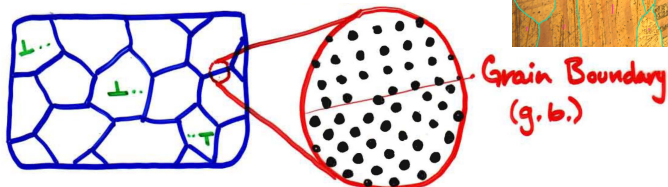
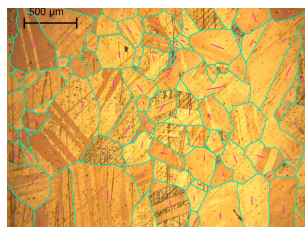
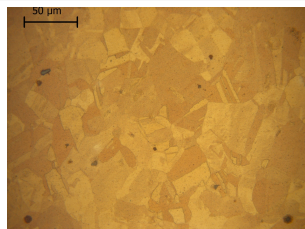
*In a heavily deformed metal, there can be as much as  $\approx 10^{12}$  cm of dislocation line per  $\text{cm}^3$  of material.*

$10^{12} \text{ cm} = 10^7 \text{ km} \approx 25 \times \text{distance to the moon}$



## MICROSTRUCTURE IN SOLIDS:

- Solids formed from solidification; this leads to many micro-crystals called grains.
- Difficult for dislocations to cross a grain boundary
- By cooling faster, smaller grains are formed; i.e., more grain boundaries.
- When a solid is heated, large grains grow at the expense of smaller grains.



## SUMMARY:

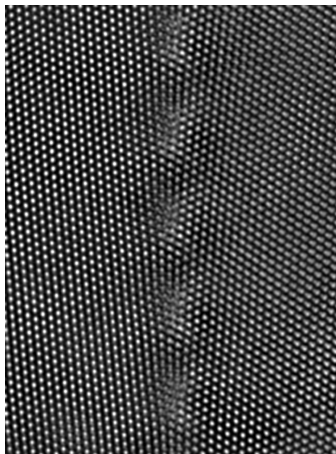


Image by: SETH T. TAYLOR, Nature Materials, 3 682 (2004).

- Stretching of atomic bonds is reversible and occurs during elastic deformation.
- Dislocation motion is irreversible and leads to permanent or plastic deformation.
- Solids are usually composed of several microscopic crystals known as grains.
- The size of grains increases (grain growth) when a solid material is exposed to higher temperatures.
- Dislocation motion is impeded by grain boundaries; hence, smaller grained metals tend to be more difficult to deform plastically.





# LAB WORKSHEET

1) Write the names of your group members below.

_____	_____
_____	_____
_____	_____

2) Write the letter of your brass sample here \_\_\_\_\_.

3) Write the hardness values you measured for your sample below.

\_\_\_\_\_

4) After consulting with the other members in your group, write the temperature at which each sample was annealed (500°C, 600°C, 700°C, or 800°C) below.

_____	_____	_____	_____
Sample A	Sample B	Sample C	Sample D

5) Which sample do you expect to have the smallest grain size, and which sample do you expect to have the largest grain size?

\_\_\_\_\_  
smallest grains  
(A, B, C, or D)

\_\_\_\_\_  
largest grains  
(A, B, C, or D)

6) On the back of this page, briefly explain how you came to your conclusions for 4) and 5).

